organic papers

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Shu-Sheng Zhang,* Jun Wan, Zheng-Zhong Peng and Sai Bi

College of Chemistry and Molecular Engineering, Qingdao University of Science and Technology, 266042 Qingdao, Shandong, People's Republic of China

Correspondence e-mail: shushzhang@126.com

Key indicators

Single-crystal X-ray study T = 293 KMean σ (C–C) = 0.004 Å R factor = 0.048 wR factor = 0.118 Data-to-parameter ratio = 15.1

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

3-(Benzotriazol-1-yl)-1-(4-chlorophenyl)-1-oxopropan-2-yl 2-chlorobenzoate

In the title molecule, $C_{22}H_{15}Cl_2N_3O_3$, the benzotriazole mean plane makes dihedral angles with the two benzene rings of 11.29 (1) and 84.90 (1)°. Weak intermolecular $C-H\cdots O$ hydrogen bonds link the molecules into chains extended along the *b* axis.

Comment

Recently, we have reported the crystal structure of 2-(1H-1,2,3-benzotriazol-1-ylmethyl)-1-benzoylethyl 4-chlorobenzoate, (II) (Wan*et al.*, 2006). As part of our ongoing search fornew benzotriazole compounds with high bioactivity, the titlecompound, (I), was synthesized. We present here its crystalstructure.



In (I) (Fig. 1), all bond lengths and angles are within normal ranges (Allen *et al.*, 1987) and comparable with those in the related compound (II). The benzotriazole mean plane makes dihedral angles with the two benzene rings, C1–C6 and C17–C22, of 11.29 (1) and 84.90 (1)°, respectively. The dihedral angle between the benzene rings is 73.63 (1)°.

In the crystal structure (Fig. 2), weak intermolecular C– H···O hydrogen bonds (Table 1) link the molecules into chains along the *b* axis. The distance of 3.728 (2) Å between the centroids of rings N1–N3/C10/C11 and C10–C15 related by the symmetry element (-x, -y, 1 - z) suggests a possible π – π interaction.

Experimental

The title compound was prepared according to the literature method of Wan *et al.* (2006). Single crystals suitable for X-ray diffraction were obtained by slow evaporation of an ethyl acetate solution at room temperature over a period of 6 d.

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Crystal data

 $C_{22}H_{15}Cl_2N_3O_3$ $M_r = 440.27$ Monoclinic, $P2_1/c$ a = 10.112 (2) Å b = 9.2772 (19) Å c = 21.998 (5) Å $\beta = 92.773$ (3)° V = 2061.2 (8) Å³

Data collection

Siemens SMART 1000 CCD area-	
detector diffractometer	
ω scans	
Absorption correction: multi-scan	
(SADABS; Sheldrick, 1996)	
$T_{\rm min} = 0.904, T_{\rm max} = 0.990$	

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_0^2) + (0.0529P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.048$	+ 0.1582P]
$wR(F^2) = 0.118$	where $P = (F_0^2 + 2F_c^2)/3$
S = 1.02	$(\Delta/\sigma)_{\rm max} < 0.001$
4081 reflections	$\Delta \rho_{\rm max} = 0.21 \text{ e } \text{\AA}^{-3}$
271 parameters	$\Delta \rho_{\rm min} = -0.26 \text{ e} \text{ Å}^{-3}$
H-atom parameters constrained	

Z = 4

 $D_x = 1.419 \text{ Mg m}^{-3}$

 $0.30 \times 0.19 \times 0.03 \text{ mm}$

11245 measured reflections 4081 independent reflections 2690 reflections with $I > 2\sigma(I)$

Mo $K\alpha$ radiation

 $\mu = 0.34 \text{ mm}^{-1}$ T = 293 (2) K

Plate, white

 $R_{\rm int} = 0.029$ $\theta_{\rm max} = 26.1^{\circ}$

Table 1

Hydrogen-bond geometry (Å, °).

$D - H \cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - H \cdot \cdot \cdot A$
C8-H8A···O1 ⁱ	0.98	2.49	3.455 (3)	170
$C9-H9A\cdots O3^n$	0.97	2.51	3.117 (3)	121

Symmetry codes: (i) $-x, y + \frac{1}{2}, -z + \frac{1}{2}$; (ii) $-x, y - \frac{1}{2}, -z + \frac{1}{2}$.

All H atoms were located in difference Fourier maps and constrained to ride on their parent atoms, with C-H = 0.93-0.98 Å and $U_{iso}(H) = 1.2U_{ed}(C)$.

Data collection: *SMART* (Siemens, 1996); cell refinement: *SAINT* (Siemens, 1996); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997a); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997a); molecular graphics: *SHELXTL* (Sheldrick, 1997b); software used to prepare material for publication: *SHELXTL*, *PARST* (Nardelli, 1995) and *PLATON* (Spek, 2003).

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Figure 1

The molecular structure of (I), showing 50% probability displacement ellipsoids and the atom-numbering scheme.



Figure 2

Packing diagram of (I), viewed down the *a* axis, showing the intermolecular hydrogen bonds as dashed lines.

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